Bioactive Nonanolide Derivatives Isolated from the Endophytic Fungus *Cytospora* sp.

Shan Lu,[†] Peng Sun,[†] Tiejun Li,[†] Tibor Kurtán,[‡] Attila Mándi,[‡] Sándor Antus,[‡] Karsten Krohn,[§] Siegfried Draeger,^{||} Barbara Schulz,^{||} Yanghua Yi,[†] Ling Li,^{*,†} and Wen Zhang^{*,†}

[†]Research Center for Marine Drugs, and Department of Pharmacology, School of Pharmacy, Second Military Medical University, 325 Guo-He Road, Shanghai 200433, P. R. China

[‡]Department of Organic Chemistry, University of Debrecen, POB 20, H-4010 Debrecen, Hungary

[§]Department of Chemistry, Universität Paderborn, Warburger Straße 100, 33098 Paderborn, Germany

^{II}Institut für Mikrobiologie, Technische Universität Braunschweig, Spielmannstraße 7, 31806 Braunschweig, Germany

S Supporting Information

ABSTRACT: Cytospolides $F-Q_{(6-17)}$ and decytospolides A and B (18 and 19), 14 unusual nonanolide derivatives, were isolated from *Cytospora* sp., an endophytic fungus from *Ilex canariensis*. The structures were elucidated by means of detailed spectroscopic analysis, chemical interconversion, and X-ray single crystal diffraction. The solution- and solid-state conformers were compared by the combination of experimental methods (X-ray, NMR) supported by DFT calculations of the conformers. Absolute configurations were assigned using the modified Mosher's method and solution- and



solid-state TDDFT ECD calculations. In an in vitro cytotoxicity assay toward the tumor cell lines of A549, HCT116, QGY, A375, and U973, the γ -lactone 17 demonstrated a potent growth inhibitory activity toward the cell line A-549, while nonanolide 16 with (2S) configuration showed the strongest activity against cell lines A-549, QGY, and U973. A cell cycle analysis indicated that compound 16 can significantly mediate G1 arrest in A549 tumor cells, confirming the important role of the C-2 methyl in the growth inhibition toward the tumor line. The discovery of an array of new nonanolides demonstrates the productivity of the fungus, and it is an example of chemical diversity, extending the nonanolide family by derivatives formed by ring cleavage, oxidation, esterification, and Michael addition.

INTRODUCTION

Secondary metabolites belonging to the family of nonanolides (synonym decanolides) are constructed of a 10-membered macrolide core with a C-9 alkyl chain. Structurally more complex nonanolides with additional rings are sometimes also included in the family.¹ The early investigations on nonanolides can be traced back to 1942 when the first example, jasmine ketolactone, was isolated as a component of the essential oil of Jasminum grandiflorum.² The structure of this jasmine ketolactone was fully elucidated in 1964.³ In 1975, diplodialide A was isolated as a steroid hydroxylase inhibitor from Diplodia pinea, representing the second report of such a structure.⁴ The discovery of a series of new nonanolides from a variety of natural sources, mainly from microorganisms,^{1,5-18} has continuously extended the knowledge on this group of metabolites. Nonanolides were also isolated from several marine invertebrates¹ and plants¹ and were found to have various biological activities, including cytotoxic,¹ phyto-toxic,^{1,5-8} antimalarial,⁹ antifungal,^{1,10,11} antibacterial,^{1,12-14} and antimicrofilament activities,¹⁵ as well as inhibitory effects against fish embryo larval development,¹⁶ AChE,¹⁷ and calmodulin-dependent cAMP phosphodiesterase.¹⁸ The broad spectrum of bioactivity and the intriguing structure of the medium-sized ring in nonanolide analogues have drawn the attention of synthetic chemists, resulting in a great number of publications. These scientific results were subsequently summarized in several reviews, $^{1,19-21}_{1,19-21}$ followed by an increasing number of reports.²²⁻⁴² The chemistry of some nonanolides has been studied intensively, and their total synthesis was reported repeatedly. Some examples include the cholesterol biosynthesis inhibitor decarestrictine, $^{1,19,20,22-25}_{1,19,21,23,26-30}$ the antimicro-filament agent microcarpalide, $^{1,19,21}_{1,19,21,23,26-30}$ and the bacterial DNA primase inhibitor Sch 642305.

During our ongoing screening for biologically active secondary metabolites from fungi,⁴³⁻⁴⁶ we recently investigated *Cytospora* sp., an endophytic fungus from *Ilex canariensis* (Aquifoliaceae, Aquifoliales), an evergreen shrub from the island of Gomera, Spain. The fungus was cultivated on biomalt agar medium. The crude extract of the culture showed pronounced antifungal activity against *Microbotryum violaceum* and moderate algicidal activity against *Chlorella fusca*. Preliminary investigation of the crude acetone extract in our laboratory led to the isolation and structural elucidation of five novel 10-membered macrolides, cytospolides A–E (1–5).⁴⁶ The skeleton of cytospolides A–E (1–5) was constructed of 15 carbons and extended the nonanolide family by a novel carbon

Received: August 23, 2011 Published: October 19, 2011 skeleton with a unique structural feature of a C-2 methyl group. Both C-2 epimers were isolated from the same fungal extract, and the epimers differed considerably in their cytotoxic activities against the A549 cell line.⁴⁶ Our investigation on the trace compounds of the crude extract has now resulted in the isolation of additional new members of this family, namely the cytospolides F-Q (6–17) and decytospolides A and B (18 and 19). Compounds 6–12 and 16 are analogues of cytospolides A–D (1–4) and cytospolide E (5), respectively. The cyclization products 13–15 and rearrangement products 17 are furan containing lactones, while the lipid ketones 18 and 19 are decarboxylated derivatives of cytospolides. Herein, we report the isolation, structural elucidation, including relative and absolute configuration, and bioactivities of these compounds.

RESULTS AND DISCUSSION

The fungus *Cytospora* sp. was cultivated on biomalt agar medium for 4 weeks and then extracted with acetone. The crude extract was fractionated on a silica gel column, followed by Sephadex LH-20 column chromatography and RP-HPLC, to yield pure cytospolides F-Q (6–17) and decytospolides A and B (18 and 19). Structures of these metabolites were elucidated by a combination of detailed spectroscopic analysis, chemical interconversion, X-ray single crystal diffraction, and comparison with data of reported cytospolides A-E (1–5).⁴⁶ The absolute configurations were determined by the modified Mosher's method and TDDFT calculations of ECD spectra, including the solid-state TDDFT ECD approach.



Cytospolide F (6) was obtained as an optically active colorless oil. Its molecular formula C15H26O5 was established by HRESIMS, indicating an additional oxygen atom in comparison to that of 4. The ¹H and ¹³C NMR spectra of 6 showed signals identical to those of 4 in the 10-membered lactone ring, except for the signals of the side chain, in which a hydroxyl group was introduced at C-13 (Tables 1 and 2). The location of 13-OH was deduced by the doublet signal of H₃-14 and confirmed by the correlations from H₃-14 to H₃-15 established by the ${}^{1}H-{}^{1}H$ COSY and HMBC spectra. The NOESY spectrum of 6 suggested the same relative configuration of the 10-membered lactone ring as that of 4. Similarly to the ECD spectrum of 4,⁴⁶ the ECD spectrum of 6 showed a negative Cotton effect (CE) at 195 nm, indicating the same (2R) absolute configuration. On the basis of the relative configuration, the absolute configuration of the lactone ring of 6 was then assigned as (2R,3R,8S,9R).

In order to determine the absolute configuration of the secondary C-13 hydroxyl group by the Mosher's method,^{47–49} diastereomeric triesters were prepared with (*R*)- and (*S*)-MTPA. As shown in Figure 1, the positive and negative $\Delta\delta$ ($\delta_{\rm S}-\delta_{\rm R}$)

values for the protons flanking the C-8 chirality center suggested (8S) absolute configuration in accordance with the CD result. In contrast, protons flanking the C-13 chirality center showed positive $\Delta\delta$ ($\delta_{\rm S} - \delta_{\rm R}$) values at both sides making the assignment ambiguous, which is presumably due to the presence of the C-8 MTPA ester group. However, 14-H had much smaller positive value (+15.4) than 10–12-H (+112.5, +141.7, +58.8, respectively), which is most likely governed by the C-13 MTPA ester group and suggested (13S) absolute configuration (Figure 1). The chemical shift difference data of MTPA diester and monoester derivatives of related derivatives 11 and 18 corroborate well our assignation (vide infra). The absolute configuration of compound 6 was thus elucidated as $(-)-(2R_3R_8S_9R_113S)$ -6.

Cytospolide G (7), an optically active colorless oil, had the molecular formula $C_{17}H_{28}O_6$ as deduced from HRESIMS. The ¹H and ¹³C NMR spectra of 7 (Tables 1 and 2) resembled those of **6**, revealing their similar structures. The presence of signals at $\delta_{\rm H} = 2.14$, $\delta_{\rm C} = 170.4$, and 20.9 in combination with the typical downfield shift of H-3 from 4.30 in **6** to 5.34 in 7 indicated the presence of an acetyl group at C-3. Detailed 2D NMR analysis, including HSQC, HMBC, ¹H–¹H COSY, and NOESY, supported the above conclusion. Acetylation of **6** and 7 gave the triacetyl derivative **20** with identical data in all respects, including MS, NMR, and $[\alpha]^{20}_{\rm D}$ (Experimental section). The structure and absolute stereochemistry of 7 was thus determined as the 3-acetyl derivative of **6** with the (-)(2R,3R,8S,9R,13S) absolute configuration.

Cytospolide H (8) was isolated as an optically active colorless oil. Its molecular formula $C_{19}H_{30}O_7$ was established by HRESIMS. Again, the ¹H and ¹³C NMR spectra of 8 (Tables 1 and 2) showed similarity to those of 7 with the exception of an additional acetyl group ($\delta_{\rm H} = 2.03$, $\delta_{\rm C} = 170.8$, and 21.4). The location of the acetyl group at C-13 was shown by the typical downfield shift of the proton signal at C-13 from 3.80 in 7 to 4.89 in 8 (Tables 1 and 2). Acetylation of 8 resulted in the same product (**20**) (Experimental Section) as that of **6** and 7. The structure of **8** was then determined as shown above with (-)-(2*R*,3*R*,8*S*,9*R*,13*S*) absolute configuration.

Cytospolides I (9) and J (10) were isolated as a 7:3 mixture that could not be separated using normal-phase column chromatography or reversed-phase HPLC techniques. However, the structural elucidation of both compounds from the mixture was possible because of the well-separated resonance signals observed in their ¹H and ¹³C NMR spectra (Tables 1 and 2). Compounds 9 and 10 had the same molecular formula of C₁₇H₂₈O₆ as established by ESIMS, ¹³C NMR, and DEPT spectra. The ¹H and ¹³C NMR spectra of both compounds showed patterns similar to that of 6 in the 10-membered lactone ring. Differences were only recognized in the substituent pattern of an acetoxyl group at the side chain. The position of the acetoxy group in 9 was obviously the same as that in 8 due to their identical ¹³C NMR data from C-10 to C-15 (Table 2). In the ¹H and ¹³C NMR spectra of 10, the appearance of a primary oxygenated carbon ($\delta_{\rm H}$ = 4.06, t, J = 6.7; $\delta_{\rm C}$ = 64.4, t) and a lack of the 14-methyl signals led the location of the acetoxyl group at C-14. The established structures were fully supported by 2D NMR experiments.

Cytospolide K (11) was also obtained as an optically active colorless oil, with the molecular formula of $C_{15}H_{26}O_4$, as deduced from HRESIMS. The ¹H and ¹³C NMR spectra of 11 were closely related to those of **6**, except that the secondary alcohol at C-8 ($\delta_{\rm H}$ = 3.65, $\delta_{\rm C}$ = 73.6/73.8) in **6** was replaced by

Table 1.	¹ H NMR	Data	for	Cytos	polides	F-I	(6 - 10)) ^a
I ubic 1.	11 141414	Dutu	101	0,000	pondes	- J '	0 10	,

no.	6	7	8	9	10
2	2.68, qd, 6.9, 3.1	2.73, qd, 7.2, 3.6	2.73, qd, 7.0, 3.1	2.70, qd, 7.0, 3.0	2.70, qd, 7.0, 3.0
3	4.30, brs	5.34, brs	5.34, brs	4.32, brs	4.32, brs
4	5.60, ov	5.59, dd, 16.2, 2.4	5.59, d, 16.1	5.62, ov	5.62, ov
5	5.60, ov	5.52, ddd, 16.2, 9.6, 4.2	5.55, ddd, 16.1, 9.2, 4.3	5.62, ov	5.62, ov
6α	2.14, m	2.09, m	2.11, m	2.16, m	2.16, m
6β	2.39, m	2.35, m	2.36, m	2.40, m	2.40, m
7α	1.99, m	1.97, m	1.98, m	2.00, m	2.00, m
7β	1.83, m	1.79, m	1.83, m	1.85, m	1.85, m
8	3.65, t, 7.3	3.64, t, 7.2	3.62, m	3.65, t, 7.0	3.65, t, 7.0
9	4.76, m	4.76, dt, 7.2, 3.6	4.75, dt, 7.2, 3.6	4.77, dt, 7.2, 3.6	4.77, dt, 7.2, 3.6
10a	1.81, m	1.78, m	1.55, m	1.73, m	1.75, m
10b	1.63, m	1.60, m	1.55, m	1.55, m	1.75, m
11a	1.42, m	2.07, m	1.33, m	1.30, m	1.32, m
11b	1.42, m	2.01, m	1.33, m	1.30, m	1.32, m
12a	1.45, m	1.45, m	1.55, m	1.45, m	1.33, m
12b	1.45, m	1.40, m	1.55, m	1.45, m	1.33, m
13a	3.80, m	3.80, m	4.89, m	4.91, m	1.62, m
13b					1.62, m
14	1.18, d, 6.0	1.18, d, 6.1	1.19, d, 6.2	1.21, d, 6.2	4.06, t, 6.7
15	1.28, d, 7.0	1.17, d, 7.0	1.18, d, 7.0	1.30, d, 7.0	1.30, d, 7.0
3-OAc		2.14, s	2.14, s		
13-OAc			2.03, s	2.04, s	
14-OAc					2.06, s
^{<i>i</i>} In CDCl ₃ , a	ssignments made by DEl	PT, ¹ H– ¹ H COSY, HSOC. HM	ABC, and NOESY.		

a methylene single ($\delta_{\rm H} = 1.50$ and 1.81; $\delta_{\rm C} = 33.5$) in 11 (Table 3). Analysis of the ¹H-¹H COSY and HMBC spectra of 11 supported the proposed planar structure. On the basis of the corresponding chirality centers of the lactone ring, the NOESY experiment suggested that 11 has the same relative configuration as 6. The ECD curve of 11 had a negative CE at 195 nm in agreement with those of 4 and 6 indicating the same (2R) absolute configuration. The absolute configuration of the three chirality centers of the lactone ring was thus assigned as (2R,3R,9S).

The absolute configuration of C-13 was determined by the modified Mosher's method.^{47–49} The negative $\Delta\delta$ ($\delta_{\rm S} - \delta_{\rm R}$) value of H-14 and the positive ones of H-8–H-12 allowed determination of the absolute configuration at C-13 as (13*S*) (see the Supporting Information, Figure S1). Moreover, the negative $\Delta\delta$ ($\delta_{\rm S} - \delta_{\rm R}$) value of 15-H and the positive ones of H-4–H₂-6 corroborated the absolute configuration determined by ECD. The absolute configuration of compound 11 was thus elucidated unambiguously as (–)-(2*R*,3*R*,9*S*,13*S*)-11.

Cytospolide L (12) was isolated as an optically active colorless oil with its molecular formula of $C_{17}H_{28}O_5$ being established by HRESIMS. The ¹H and ¹³C NMR spectra of 12 displayed a great similarity to those of 11, except for the presence of an additional acetyl group ($\delta_H = 2.02$, s; $\delta_C = 170.8$, s



Figure 1. $\Delta\delta$ (δ_{s} - δ_{R}) values (in Hz) for the MTPA esters of **6**.

and 21.4, q) (Table 3). The location of the acetyl group at C-13 was assured by the downfield shift of the H-13 ($\delta_{\rm H}$ = 4.87 in 12 and 3.78 in 11), so that compound 12 was determined as the 13-acetyl derivative of 11. The long-range correlation from H-13 to the acetyl carbonyl atom ($\delta_{\rm C}$ = 170.8) in the HMBC spectrum further confirmed the proposed structure. Acetylation of 11 and 12 resulted in the same product 21 (Experimental Section). The absolute stereochemistry of compound 12 was thus determined to be the same as that of 11.

Interestingly, carbon atoms of C-8, C-12, C-13, and C-14 of compounds 6-9 were observed to resonate as two signals when at least one of the C-8 and C-13 carbons has a free hydroxyl group. The two series of signals may be due to two different intramolecular hydrogen-bonding interactions. These duplicated signals change into single ones when intramolecular hydrogen bonds are broken, i.e., if one of the oxygenated groups at C-8 and C-13 was absent (e.g., in $1-5^{46}$ and 10-12 see Table 2) or none of the C-8 and C-13 substituents was a free hydroxyl group (e.g., acetylating 6-8 to 20).

Cytospolide M (13) was isolated as optically active colorless crystals. The HRESIMS of 13 established the molecular formula $C_{15}H_{26}O_5$, indicating three double bond equivalents. The IR absorption showed the presence of hydroxy (3346 cm⁻¹) and ester carbonyl (1731 cm⁻¹) groups. The ¹³C NMR and DEPT spectra revealed one sp² carbon of an ester carbonyl atom at lower field and fourteen sp³ carbons (1 × CH, 6 × CH₂, 2 × CH₃, and 5 × OCH at higher field, which were assigned to their corresponding proton signals by HSQC experiments (Table 4). Obviously, there are two rings present in the structure based on the established molecular formula and the relevant NMR data.

The planar structure and $(2R^*, 3S^*, 4R^*, 5R^*, 8S^*, 9R^*)$ relative stereochemistry of **13** were determined by both NMR (Figure 2 and 3) and single-crystal X-ray analysis (Figure 4), the geometry of the latter was then used to determine the absolute

Table 2. ¹³C NMR Data for Cytospolides F-J (6–10)^{*a*}

	4	7	0	0	10
110.	0	/	8	9	10
1	174.3/174.4, C	172.4, C	172.3, C	174.3, C	174.4, C
2	46.7, CH	44.9/45.0, CH	45.0, CH	46.7, CH	46.7, CH
3	72.0 CH	72.5/72.6, CH	72.5, CH	72.0, CH	72.0, CH
4	133.2, CH	128.9, CH	128.9, CH	133.2, CH	133.2, CH
5	127.5, CH	129.5, CH	129.6, CH	127.5, CH	127.5, CH
6	28.6/28.9, CH ₂	29.4, CH ₂	29.3, CH ₂	28.4, CH ₂	28.5, CH ₂
7	38.6, CH ₂	38.8, CH ₂	38.6, CH ₂	38.5, CH ₂	38.6, CH ₂
8	73.6/73.8, CH	73.6/73.8, CH	73.6/73.9, CH	73.6/73.8, CH	73.4, CH
9	77.6, CH	77.4/77.5, CH	77.3, CH	77.4, CH	77.6, CH
10	31.5/32.1, CH ₂	32.1, CH ₂	32.0, CH ₂	32.0, CH ₂	32.1, CH ₂
11	20.3/20.8, CH ₂	20.7, CH ₂	20.2/20.6, CH ₂	20.2/20.6, CH ₂	24.3, CH ₂
12	38.8/38.9, CH ₂	38.5/38.6, CH ₂	35.7/35.8, CH ₂	35.7/35.8, CH ₂	25.9, CH ₂
13	67.5/68.0, CH	67.5/68.1, CH	70.6/70.8, CH	70.6/70.7, CH	31.7, CH ₂
14	23.5/23.6, CH ₃	23.5/23.7, CH ₃	19.9/20.1, CH ₃	19.9/20.1, CH ₃	64.4, CH ₂
15	12.5, CH ₃	12.2, CH ₃	12.2, CH ₃	12.4, CH ₃	12.4, CH ₃
3-OAc		170.4, C	170.4, C		
		20.9, CH ₃	20.9, CH ₃		
13-OAc			170.8, C	170.8/170.9, C	
			21.4, CH ₃	21.4, CH ₃	
14-OAc					171.3, C
					21.0, CH ₃

"In CDCl₃, assignments made by DEPT, ¹H-¹H COSY, HSQC, and HMBC.

Table 3. NMR Data for Cytospolides K and L $(11 \text{ and } 12)^a$

	11		12		
no.	$\delta_{ m H\prime}$ m, J (Hz)	$\delta_{ m C'}$ m	$\delta_{ m H}$, m, J (Hz)	δ_{C} m	
1		174.4, C		174.3, C	
2	2.69, qd, 6.9, 3.1	47.1, CH	2.68, qd, 6.8, 3.3	47.1, CH	
3	4.30, brs	72.8, CH	4.31, brs	72.8, CH	
4	5.62, d, 17.4	133.4, CH	5.62, d, 17.5	133.4, CH	
5	5.57 (ddd, 17.4, 9.5, 3.7)	127.3, CH	5.57, ddd (17.5, 9.0, 3.0)	127.4, CH	
6α	1.95, m	32.7, CH ₂	1.95, m	32.6, CH ₂	
6β	2.30, m		2.30, m		
7α	1.84, m	28.1, CH ₂	1.84, m	28.0, CH ₂	
7β	1.45, m		1.44, m		
8α	1.50, m	33.5, CH ₂	1.47, m	33.4, CH ₂	
8β	1.81, m		1.78, m		
9	4.75, q, 6.0	76.2, CH	4.74, q, 6.0	76.0, CH	
10a	1.55, m	35.8, CH ₂	1.55, m	35.5, CH ₂	
10b	1.45, m		1.47, m		
11a	1.40, m	21.5, CH ₂	1.31, m	21.1, CH ₂	
11b	1.40, m		1.31, m		
12a	1.45, m	38.9, CH ₂	1.55, m	35.6, CH ₂	
12b	1.45, m		1.55, m		
13	3.78, m	67.9, CH	4.87, m	70.7, CH	
14	1.18, d, 6.1	23.6, CH ₃	1.19, d, 6.5	19.9, CH ₃	
15	1.28, d, 7.0	12.5, CH ₃	1.28, d, 7.0	12.5, CH ₃	
13-OAc			2.02, s	170.8, C	
				21.4, CH ₃	

^{*a*}In CDCl₃, assignments made by DEPT, ¹H-¹H COSY, HSQC, HMBC, and NOESY.

configuration of **13** by the solid-state TDDFT ECD approach.⁵⁰ In this method, the DFT-optimized X-ray structure is used as input for TDDFT ECD calculations,^{51,52} and the resultant computed ECD spectrum is compared with the experimental solid-state ECD measured as a KCl disk (Figure 5). The solution- and solid-state ECD spectra of **13** showed a negative band at 214 nm and a positive one below 200 nm. Their good agreement suggested that the solid-state X-ray conformer is also

prevalent in solution. This was confirmed by a conformational analysis on the truncated 9-methyl model compound of 13, which showed that the solid-state conformer had 94.8% population (see the Supporting Information, Figure S2). The ECD spectra calculated for the optimized (2*R*,3*S*,4*R*,5*R*,8*S*,9*R*) enantiomer of the X-ray structure reproduced well the experimental solid-state ECD, allowing the determination of absolute configuration.

Article

Table 4. NMR Data for Cytospolides M, N, and Q $(13, 14, and 17)^a$

	13		14		17	
	$\delta_{\rm H}$, m, J (Hz)	$\delta_{\rm C}$, m	$\delta_{\rm H}$, m, J (Hz)	$\delta_{\rm C}$, m	$\delta_{\rm H}$, m, J (Hz)	$\delta_{ m C'}$ m
1		175.1, C		175.3, C		177.9, C
2	2.75, qd, 6.9, 2.5	45.6, CH	2.75, qd, 7.0, 2.3	45.5, CH	2.80, m, 7.4	39.2, CH
3	3.88, brs	81.9, CH	3.88, brs	81.8, CH	4.51, dd, 7.0, 3.6	71.1, CH
4	3.55, d, 9.1	76.5, CH	3.53, t, 7.2	76.4, CH	4.12, dd, 8.0, 3.6	85.2, CH
5	4.10, dd, 9.1, 7.0	80.6, CH	4.10, dd, 8.4, 7.2	80.6, CH	3.94, dt, 7.5, 4.0	78.5, CH
6α	2.08, m	29.5, CH ₂	2.08, m	29.5, CH ₂	2.09, m	28.9, CH ₂
6β	2.00, m		2.00, m		1.95, m	
7α	1.87, m	23.4, CH ₂	1.86, m	23.5, CH ₂	1.83, m	23.1, CH ₂
7β	1.84, m		1.84, m		1.94, m	
8	4.03, dt, 8.3, 3.8	81.2, CH	4.02, dt, 7.9, 3.8	81.1, CH	3.90, dt, 8.0, 3.0	83.4, CH
9	5.36, dt, 9.1, 3.8	75.0, CH	5.34, br d, 8.8	74.9, CH	3.82, dt, 7.8, 3.6	71.5, CH
10a	1.54, m	29.1, CH ₂	1.54, m	28.9, CH ₂	1.38, m	33.1, CH ₂
10b	1.45, m		1.45, m		1.38, m	
11a	1.35, m	25.6, CH ₂	1.43, m	25.6, CH ₂	1.50, m	25.6, CH ₂
11b	1.35, m		1.43, m		1.33, m	
12a	1.32, m	31.6, CH ₂	1.43, m	25.7, CH ₂	1.30, m	31.8, CH ₂
12b	1.32, m		1.43, m		1.30, m	
13a	1.31, m	22.4, CH ₂	1.64, m	28.4, CH ₂	1.32, m	22.6, CH ₂
13b	1.31, m		1.64, m		1.32, m	
14	0.90, t, 6.6	14.0, CH ₃	4.06, t, 6.8	64.3, CH ₂	0.90, t, 6.5	14.0, CH ₃
15	1.30, d, 6.9	14.6, CH ₃	1.29, d, 6.9	14.6, CH ₃	1.27, d,7.5	8.8, CH ₃
14-OAc			2.04, s	171.2, C		
				210 CH		

^{*a*}In CDCl₃, assignments made by DEPT, ¹H–¹H COSY, HSQC, HMBC, and NOESY.

Cytospolide N (14) was isolated as an optically active colorless gum. Its molecular formula $C_{17}H_{28}O_7$ was established by HRESIMS. The ¹H and ¹³C NMR spectra of 14 (Table 4) were almost identical to those of 13, except for replacement of signals for the 14-methyl group ($\delta_C = 14.0$; $\delta_H = 0.9$) in 13 by signals of an oxygenated methylene group ($\delta_C = 64.3$; $\delta_H = 4.06$) in 14. Moreover, signals of an acetyl group ($\delta_C = 171.2$, 21.0 and $\delta_H = 2.04$) indicated that compound 14 is the 14-acetoxy derivative of 13. A detailed 2D NMR analysis, including ¹H–¹H COSY, HMBC, and NOESY, supported the established structure and relative configuration. The absolute configuration at C-2 of 14 was proven to be the same as that of 13 by its ECD spectrum, showing a similar negative band above 200 nm,



Figure 2. $^{1}\mathrm{H}\mathrm{-}^{1}\mathrm{H}$ COSY (bond) and selected HMBC (arrow) correlations of 13.



Figure 3. Key NOESY correlations of 13 indicated on the lowest energy DFT solution conformer of the truncated 9-methyl derivative.

which afforded the absolute configuration as (2*R*,3*S*,4*R*,-5*R*,8*S*,9*R*)-14.

Cytospolide O (15), an optically active colorless oil, has the molecular formula C15H24O4 as deduced from HRESIMS. A comparison of the ¹H and ¹³C NMR spectra of 15 with those of 13 revealed similarity. However, the 3-OH and 4-OH groups of 13 were replaced by a carbonyl oxygen and a proton, respectively (Table 5). The C-4 methylene group was assigned by the proton spin system from H2-4 to H3-14 as deduced from the ${}^{1}H-{}^{1}H$ COSY spectrum (Figure 6). The C-3 ketone carbonyl group was confirmed by the observation of long-range correlation of both H-2 and H₃-15 with C-1 and C-3, H-4 with C-2 and C-3 in the HMBC spectrum. The diagnostic NOE cross peaks between H-4 α and H-2, H-6 α , and H-7 α , and between H-7 α and H-10b indicated an α configuration of these protons, and H-6 β , H-8 and H-9 were therefore identified as β protons. The observed NOE effects between H-8 and both H-6 β and H-9 supported the established (2R*,5R*,8S*,9R*)



Figure 4. Single-crystal X-ray structure of 13 (ORTEP drawing).



Figure 5. Experimental solution (MeCN) and solid-state (KCl) ECD spectra of cytospolide M (13) compared with the TDDFT ECD spectra calculated for the optimized X-ray geometry of the (2R,3S,4R,5R,8S,9R) enantiomer of 13. Bars represent the calculated rotatory strength (*R*) obtained with the PBE0/TZVP method.

Table 5. NMR Data for Cytospolides O and P $(15 \text{ and } 16)^a$

	15		16	
no.	$\delta_{\rm H^{\rm J}}$ m, J (Hz)	δ_{C} m	$\delta_{\rm H}$, m, J (Hz)	$\delta_{C'}$ m
1		170.6, C		170.3, C
2	3.73, q, 6.7	57.2, CH	3.46, q, 7.0	55.7, CH
3		203.6, C		201.2, C
4α	3.14, dd, 11.7, 4.4	47.1, CH ₂	2.59, dd, 15.0, 5.3	41.8, CH ₂
4β	2.34, t, 11.4		3.13, dd, 15.0, 11.0	
5	4.56, m	77.2, CH	5.38, m	69.9, CH
6α	1.78, m	30.9, CH ₂	2.05, m	26.8, CH ₂
6β	2.00, m		1.38, m	
7α	2.10, m	24.0, CH ₂	1.64, m	28.8, CH ₂
7β	1.93, m		1.65, m	
8	4.07, dt, 7.0, 3.8	80.8, CH	3.55, br t, 8.8	73.5, CH
9	5.33, dt, 9.6, 3.6	75.9, CH	4.89, dt, 8.5, 3.0	79.4, CH
10a	1.60, m	29.2, CH ₂	1.83, m	32.7, CH ₂
10b	1.47, m		1.57, m	
11a	1.47, m	25.6, CH ₂	1.28, m	24.5, CH ₂
11b	1.38, m		1.28, m	
12a	1.33, m	31.6, CH ₂	1.28, m	31.5, CH ₂
12b	1.33, m		1.28, m	
13a	1.30, m	22.5, CH ₂	1.28, m	22.4, CH ₂
13b	1.30, m		1.28, m	
14	0.90, t, 7.0	14.0, CH ₃	0.88, t, 6.8	13.9, CH ₃
15	1.26, d, 6.7	10.4, CH ₃	1.33, d, 7.0	11.2, CH ₃
5-OAc			2.01, s	170.1, C
				21.1. CH ₂

^{*a*}In CDCl₃, assignments made by DEPT, ¹H-¹H COSY, HSQC, HMBC, and NOESY.



Figure 6. $^{1}\text{H}\text{-}^{1}\text{H}$ COSY (bond) and selected HMBC (arrow) correlations of 15.

relative stereochemistry (see the Supporting Information, Figure S3).

For the configurational assignment of 15, the solution TDDFT ECD protocol was pursued. The solution ECD spectra of 15 showed a positive $n \rightarrow \pi^*$ band around 300 nm and a

negative band below 250 nm. The conformational analysis of the truncated C-9 methyl derivative of **15** afforded three conformers with 79.0%, 17.1% and 3.9% populations (see Supporting Information, Figure S4). In the two high-energy conformers, the C-3 carbonyl group was rotated compared to that of the lowest-energy one. The ECD spectrum calculated for the (2R,5R,8S,9R) enantiomer of the solution conformers of truncated **15** reproduced well the experimental solution ECD (see Supporting Information, Figure S5) curve of **15**, allowing determination of the absolute configuration as (2R,5R,8S,9R).

Cytospolide P (16) was obtained as an optically active colorless crystal. The HRESIMS showed the molecular formula of $C_{17}H_{28}O_{67}$ indicating four double bond equivalents. The presence of a ketone and two ester functionalities in the molecule was deduced from the IR absorption (1736, 1708 cm⁻¹) and the ¹³C NMR and DEPT spectra ($\delta_{\rm C} = 201.2$, 170.1, and 170.3) (Table 5), accounting for three degrees of unsaturation. The remaining double bond equivalent was attributed to the presence of the lactone ring. The detailed NMR (Figures 7 and S6) and single crystal X-ray diffraction



Figure 7. $^{1}\mathrm{H}\mathrm{-}^{1}\mathrm{H}$ COSY (bond) and selected HMBC (arrow) correlations of 16.

(Figure 8) analysis of cytospolide P (16) afforded its planar structure and $(2S^*,5R^*,8S^*,9R^*)$ relative configuration.



Figure 8. Single-crystal X-ray structure of 16 (ORTEP drawing).

Although 15 and 16 had the same β -keto lactone chromophore, the characteristic $n \rightarrow \pi^*$ CE of 16 was negative $(\Delta \varepsilon = -0.88 \text{ at } 290 \text{ nm})$, opposite to that of 15 $(\Delta \varepsilon = +1.14 \text{ at } 296 \text{ nm})$, while their ECD bands above 200 nm (lactone $n \rightarrow \pi^*$) exhibited the same positive sign. For the configurational assignment of 16, the TDDFT ECD spectra were calculated for the optimized (2S,5R,8S,9R) enantiomer of the X-ray geometry. Based on the good agreement with the experimental solid-state ECD measured as KCl disk, the absolute configuration was deduced as (2S,5R,8S,9R). Thus, 16 has an opposite absolute configuration at C-2 compared to that of 15, which was responsible for the opposite CE of the ketone $n \rightarrow \pi^*$ ECD transition. A conformational analysis of 16 was also performed on a truncated C-9 methyl model compound, which showed that the solid-state conformer with equatorial 8-OH is the most abundant conformer represented by six slightly different conformers with a total population of 58.3% (see the Supporting Information, Figure S7). The 8-OH_{ax} conformer was represented by four conformers with 38.4% of the total population. In accordance with the result of the conformational analysis, the measured solution and solid-state ECD spectra were quite similar (see the Supporting Information, Figure S8).

Cytospolide Q (17) was isolated as an optically active colorless gum. Its molecular formula $C_{15}H_{26}O_5$ was established by HRESIMS. The IR absorption showed absorptions of a hydroxy (3435 cm⁻¹) and an ester carbonyl (1768 cm⁻¹) group. This was in agreement with the observation of an ester carbonyl and fourteen sp³ carbons (1 × CH, 6 × CH₂, 2 × CH₃, and 5 × OCH) in the ¹³C NMR and DEPT spectra (Table 4). Three degrees of double bond equivalence were attributed to one carbonyl group and two rings in the molecule. The proton spin system from H₃-15 to H₃-14 was readily identified by ¹H⁻¹H COSY, and the planar structure was then established by the HMBC correlations as shown in Figure 9.



Figure 9. $^{1}H^{-1}H$ COSY (bond) and selected HMBC (arrow) correlations of 17.

The relative stereochemistry of 17 was elucidated on the basis of detailed analysis on its NOE data. Concerning the lactone ring (A), the NOE effect between H-4 and H₃-15 indicated a β configuration of the protons while the NOE effect between H-3 and H-5 indicated their α orientation. As for the tetrahydrofuran ring (B), the observed NOE effects between H-4 and H-7 β , and between H-5 and H-8 led to the assignment of H-7 β as a β proton and H-5 and H-8 as α protons, respectively. Ring A is attached to ring B in a β position (see the Supporting Information, Figure S9). The relative configuration of H-9 was deduced from the homo- and heteronuclear ³J couplings.⁵³ The small coupling constant of ${}^{3}J_{H8,H9} = 3.0$ Hz indicated a gauche arrangement of H-8 and H-9, and the large coupling constant of ${}^{3}J_{C7,H9} = 11.5$ Hz, measured by means of J-modulated HMBC experiments,⁵⁴ indicated antiperiplanar orientation of C-7 and H-9, which finally led to $(2R^*, 3S^*, 4S^*, 5R^*, 8S^*, 9R^*)$ relative configuration.

A conformational analysis of the solution conformers of truncated 17 confirmed the NMR coupling pattern. In the lowest-energy conformer (67.7% population) the $\omega_{C7,C8,C9,H9}$ torsional angle was 174.2°, while the $\omega_{H8,C8,C9,H9}$ was found to be -60.7° , corroborating the values of experimental homonuclear and heteronuclear coupling constants (see the Supporting Information, Figure S10). These torsional angles have similar values in the higher-energy conformers as well, since they only differed in the orientation of their OH groups. The 9-OH was hydrogen-bonded to the oxygen of the 3-OH in conformer A, B, D with 67.7%, 12.8%, and 5.6% populations, respectively, and the 3-OH was hydrogen-bonded to the oxygen of 9-OH in two minor conformers with 8.4% and 2.3% populations.

The structure and relative stereochemistry of 17 were further confirmed by a single-crystal X-ray analysis (Figure 10). The X-ray diffraction also established the absolute stereochemistry



Figure 10. Single-crystal X-ray structure of 17 (ORTEP drawing).

of 17 as depicted in Figure 10 with the convincing absolute structure parameter 0.1 (2).

Decytospolide A (18) was isolated as an optically active colorless oil. Its molecular formula $C_{14}H_{26}O_3$ was established by HRESIMS. The absence of the ester carbonyl signal in the ¹³C NMR spectrum suggested a different structure than that of the above lactones (Table 6). Analysis of ¹H–¹H COSY data

Table 6. NMR Data for Decytospolides A and B (18 and 19)^{*a*}

	18		19	
no.	$\delta_{ m H\prime}$ m, J (Hz)	$\delta_{\rm C}$, m	$\delta_{\rm H}$, m, J (Hz)	$\delta_{\rm C}$ m
2	2.49, m	37.1, CH ₂	2.48, m	37.2, CH ₂
3		210.0, C		209.9, C
4a	2.66; dd, 15.0, 7.8;	48.4, CH ₂	2.68; dd, 15.2, 7.9;	48.2, CH ₂
4b	2.39, dd, 15.0, 4.8		2.38, dd, 15.2, 4.2	
5	3.74, m	74.1, CH	3.77, m	74.2, CH
6α	1.38, m	31.3, CH ₂	1.40, m	30.8, CH ₂
6β	1.78, m		1.75, m	
7α	2.08, m	32.9, CH ₂	2.14, m	29.4, CH ₂
7β	1.46, m		1.45, m	
8	3.26, dt, 10.2, 4.8	70.5, CH	4.45, dt, 10.0, 4.6	72.1, CH
9	3.04, dt, 9.0, 2.4	82.1, CH	3.23, dt, 9.3, 2.4	79.3, CH
10a	1.79, m	31.9, CH ₂	1.51, m	31.9, CH ₂
10b	1.36, m		1.32, m	
11a	1.47, m	25.0, CH ₂	1.40, m	24.8, CH ₂
11b	1.29, m		1.25, m	
12a	1.30, m	31.8, CH ₂	1.25, m	31.7, CH ₂
12b	1.30, m		1.25, m	
13a	1.30, m	22.6, CH ₂	1.28, m	22.6, CH ₂
13b	1.30, m		1.28, m	
14	0.89, t, 6.9	14.0, CH ₃	0.87, t, 6.8	14.0, CH ₃
15	1.05, t, 7.2	7.5, CH ₃	1.04, t, 7.3	7.5, CH ₃
8-OAc			2.04, s	170.3, C
				21.2, CH ₃
^a In CI)Cl., assignments	made by I	DEPT. $^{1}H-^{1}H$ CO	DSY. HSOC

"In $CDCl_3$, assignments made by DEPT, 'H–'H COSY, HSQC, HMBC and NOESY.

revealed two proton spin systems, i.e., H_3 -15 to H_2 -2, and H_2 -4 to H_3 -14. The HMBC correlation from protons H_2 -2 and H_2 -4 to C-3 led the connection of the two proton sequence through a ketone carbonyl. The formation of a pyran ring was apparent due to the diagnostic long-range correlation from H-5 to C-9, giving the planar structure of **18** (Figure 11). The C-5 and C-9 substituents of the furan ring had *cis* orientation, since a NOE effect was observed between the 1,3-diaxial protons of H-5 and H-9. H-8 had *trans*-diaxial relationship with H-9 due to the large coupling constant (${}^{3}J_{H8,H9} = 10.2 \text{ Hz}$) and showed a NOE effect with H-10b ($\delta_{\rm H} = 1.36$) (see the Supporting Information, Figure S11) allowing the assignment of the relative configuration. The absolute configuration of the secondary



Figure 11. $^1\text{H}{-}^1\text{H}$ COSY (bond) and selected HMBC (arrow) correlations of 18.

8-OH was determined as (8S) by the modified Mosher's method (see Supporting Information, Figure S12),^{46–49} which in the



Figure 12. Effect of cytospolides E and Q (5, 16) on cell cycle in A549 and QGY cells. Data were obtained by densitometry measurements (n = 3).

knowledge of the relative configuration afforded the absolute configuration of **18** as (*SR*,*8S*,*9R*).

Decytospolide B (19) was isolated as an optically active colorless oil. ¹H and ¹³C NMR spectra greatly resembled those of 18 except for the presence of an additional acetyl group ($\delta_{\rm H} = 2.04$, s; $\delta_{\rm C} = 170.3$, s and 21.2, q). This conclusion was in agreement with its molecular formula of C₁₆H₂₈O₄ as deduced from HRESIMS, showing 42 mass units more than 18. The location of the acetyl group at C-8 was assured by the typical downfield shift of the H-8 ($\delta_{\rm H} = 4.45$ in 19 and 3.26 in 18) (Table 6). The long-range correlation from H-8 to the acetyl carbonyl atom ($\delta_{\rm C} = 170.3$) in the HMBC spectrum further confirmed the proposed structure. The acetylation product of 18 was identical with 19, which served as further evidence for the elucidation of the planar structure and the absolute stereochemistry. Compound 19 was thus determined as the 8-acetate derivative of 18.

The discovery of an array of new nonanolides demonstrates the productivity of the fungus and it is an example of chemical diversity, extending the nonanolide family by derivatives formed by ring cleavage, oxidation, esterification and Michael addition. Cytospolides A-L(1-12) may biogenetically derive from precursor A by a dehydration involving the 5-OH, followed by oxidation at C-8, C-13, and/or C-14. Further oxidation of OH-3 would result in the reactive $\alpha_{,\beta}$ -unsaturated ketone precursor B, the C-3 carbonyl and $\Delta^{4,5}$ bond of which can be found separately in the reported nanolides. An intermolecular Michael addition to the β carbon of the $\alpha_{,\beta}$ unsaturated ketone and the acetylation of 5-OH may produce cytospolide P (16) from precursor **B**, while an intramolecular oxa-Michael addition of 8-OH to the $\alpha_{,\beta}$ -unsaturated ketone of precursor **B** may give cytospolide O (15). The subsequent oxidation of C-4 and reduction of C-3 carbonyl of cytospolide O (15) may produce cytospolides M and N (13, 14). Consequently, a transesterification of the lactone acyl group with 4-OH of cytospolide M (13) would construct the γ -lactone moiety of cytospolide Q (17). The two decytospolides A and B (18, 19) may be produced from precursor B by an initiation of a ring cleavage on the lactone bond, followed by intramolecular oxa-Michael addition of 9-OH to the α_{β} -unsaturated ketone and decarboxylation of the β -keto carboxylic acid derivative (Scheme 1).

The cytotoxic activity of compounds **6–19** toward the tumor cell lines of A549, HCT116, QGY, A375, and U973 was evaluated in vitro (Table 7). It had been reported that the absolute configuration of C-2 has an effect on the cytotoxic activity; e.g. cytospolide D (4) with (2*R*) configuration showed no activity against A-549 cell line at 50 μ g/mL, while its C-2 epimer, cytospolide E (**5**), displayed modest cytotoxic activity with an IC₅₀ value of 7.09 μ g/mL.⁴⁶ In accordance, **16** with (2*S*) configuration showed the strongest activity against cell lines A-549, QGY and U973.

Furan containing nonanolides 13, 14, 15, and 17 also showed considerably cytotoxic activity against A-549 cells. The free hydroxyl groups at C-3 and C-4 seem to play an important role for the cytotoxicity, since the inhibition remarkably decreased in compound 15 lacking the two secondary hydroxyl

Scheme 1. Proposed Biogenetic Connection of Isolated Cytospolides A-Q (1-17) and Decytospolides A and B (18 and 19)



9706

Table 7. Cytotoxic Assays toward Different Tumor Cell Lines^{a,b}

no.	A549	HCT116	QGY	A375	U937
6	_	-	_	_	-
7	-	_	-	-	-
8	-	_	-	-	-
9	-	-	-	-	-
10	-	_	-	-	-
11	20.47	_	-	-	-
12	-	_	-	-	-
13	12.29	_	-	-	-
14	16.55	_	-	-	-
15	39.17	_	-	-	-
16	2.05	_	15.82	-	28.26
17	10.55	_	-	-	-
18	-	_	-	-	-
19	14.79	_	46.79	-	-
adriamycin	0.44	0.45	0.28	0.44	0.23
^a IC ₅₀ (μg/mL). $^{b}-: IC_{50}$	$> 50 \ \mu g/mL$			

groups. The activity of decytospolide A (18) increased dramatically when the 8-OH was acetylated.

A cell cycle analysis was executed to further evaluate the tumor cell growth inhibition of the two potent compounds, cytospolides E and Q (5, 16), both having (2S) absolute configuration. Both compounds can significantly mediate G1 arrest in A549 tumor cells at 1 μ g/mL. The percentage of cells at the G1 phase was increased from 58.8% to 82.5% and 83.1% (P < 0.01), while those at the S phase were decreased from 30.3% to 11.6% and 10.2% (P < 0.01), respectively. Cytospolide E (5) was not active toward QGY cells, whereas cytospolide P (16) displayed a significant effect (10 μ g/mL), increasing the percentage at G1 phase from 56.9% to 72.0% (P < 0.01) and decreasing the percentage of cells at S phase from 30.3% to 11.6% (P < 0.01), respectively (Figure 12).

EXPERIMENTAL SECTION

General Experimental Procedures. Precoated silica gel plates were used for analytical thin-layer chromatography (TLC). Spots were detected on TLC under UV or by heating after spraying with 0.5 mL of anisaldehyde in 50 mL of HOAc and 1 mL of H₂SO₄. TLC R_f values are reported. The NMR spectra were recorded at 293K. Chemical shifts are reported in parts per million (δ), using the residual CHCl₃ signal ($\delta_{\rm H}$ 7.26 ppm) as an internal standard for ¹H NMR, and CDCl₃ $(\delta_{\rm C}$ 77.0 ppm) for ¹³C NMR; coupling constant (J) in Hz. ¹H and ¹³C NMR assignments were supported by ¹H-¹H COSY, HMQC, HMBC, and NOESY experiments. ³J_{C,H} couplings were measured by means of pulsed field gradient HMBC spectra, recorded by varying J-refocusing time τ between 0.04 and 0.16 s (10 ms interval), corresponding to $J = 1/(2\tau) = 3.1-12.5$ Hz. ${}^{3}J_{C,H}$ values were estimated with least-squares sinusoidal fit of the experimental crosspeak intensities as a function of $J^{.54}$. The following abbreviations are used to describe spin multiplicity: s = singlet, d = doublet, t = triplet, q= quartet, m = multiplet, brs = broad singlet, dd = doublet of doublets, ddd = doublet of doublets of doublets, dt = doublet of triplets, qd = quartet of doublets, ov = overlapped signals. Infrared spectra were recorded in thin polymer films. For the protocol of the solid-state circular dichroism (CD) protocol, see ref50. The mass spectra and high-resolution mass spectra were performed on a Q-TOF Micro mass spectrometer, resolution 5000. An isopropyl alcohol solution of sodium iodide (2 mg/mL) was used as a reference compound. The X-ray diffraction study was carried out using Cu K α radiation (λ = 1.542178 Å). Semi-preparative RP-HPLC was performed using a refractive index detector and a column with 5 μ m, 250 \times 10 mm size.

Culture, Extraction, and Isolation. The endophytic fungus Cytospora sp., internal strain No. ZW02, was isolated following surface sterilization from Ilex canariensis, a woody, evergreen shrub from Gomera, and was cultivated on biomalt (5% w/v) solid agar medium at room temperature for 28 days.^{43–45} The culture medium was then extracted with acetone to afford a residue (9.98 g) after removal of the solvent under reduced pressure. The extract was subjected to column chromatography (CC) on silica gel, eluted with a gradient of CH₂Cl₂ in acetone (100:1, 80:20, 50:50, 30:70, 1:100 v/v), to give 17 subfractions. Fractions 3 and 4 were first purified on silica gel CC (400-500 mesh, CH₂Cl₂/acetone, 100:1) and then fractionated by semi-preparative RP-HPLC (MeOH/H2O, 8:2, 1.5 mL/min) to give 16 (1.7 mg, 15.8 min, yield 0.17‰), 15 (6.3 mg, 20.3 min, yield 2.03‰), and 19 (1.5 mg, 27.4 min, yield 2.74‰), respectively. Fractions 8-10 were first purified on silica gel CC (400-500 mesh, CHCl₃/MeOH, 50:1), followed by a purification of semi-preparative RP-HPLC: fraction 8 afforded 12 (1.7 mg, yield 0.17‰) at 27.7 min with the elution of MeOH/H₂O (8:2, 1.0 mL/min); fractions 9 and 10 yielded pure 7 (5.4 mg, yield 0.54‰), 13 (9.4 mg, yield 0.94‰), 14 (1.0 mg, yield 0.10‰), and 18 (8.1 mg, yield 0.81‰) at 23.2, 18.3, 19.9, and 29.7 min, respectively, eluted with MeOH/H $_2O$ (7:3, 1.5 mL/min). Fraction 11 gave 6 (7.7 mg, yield 0.77‰) and 8 (0.9 mg, yield 0.09‰) by a simple fractionation on silica gel CC (400-500 mesh, CHCl₃/MeOH, 30:1). Fractions 14, 15, and 16 were purified on silica gel CC (400-500 mesh, CHCl₃/MeOH, 20:1), and then separated by semi-preparative HPLC, eluted with MeOH/H2O (7:3, 1.5 mL/min) to yield pure 11 (6.6 mg, yield 0.66‰), 17 (4.5 mg, yield 0.45‰) and the mixture products 9/10 (6.1 mg, yield 0.61‰), at 19.6, 18.6, and 14.9 min, respectively.

Cytospolide F (6): colorless oil; $R_f = 0.38$ (CHCl₃/MeOH 10:1); $[\alpha]^{20}_{D} = -72.7$ (*c* 0.08, CHCl₃); CD (CH₃CN, *c* 2.0 × 10⁻⁴) λ_{max} ($\Delta \varepsilon$) = 195 (-11.48) nm; UV (CH₃CN) λ_{max} (ε) 219 (2449) nm; IR (film) ν_{max} = 3415, 3004, 2920, 1716, 1656, 1436, 1410, 1317, 1185, 1022, 953, 903, 707 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Tables 1 and 2; HRMS (ESI) [M + Na]⁺ m/z 309.1675 (calcd for C₁₅H₂₆O₅Na, 309.1678).

Cytospolide G (7:). colorless oil; $R_f = 0.79$ (CHCl₃/MeOH 10:1); $[\alpha]^{20}_{D} = -53.7$ (*c* 0.05, CHCl₃); UV (CH₃CN) λ_{max} (ε) 221 (1797) nm; IR (film) $\nu_{max} = 3406$, 2930, 2857, 1737, 1670, 1457, 1375, 1240, 1180, 1106, 1065, 1024, 990, 943 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Tables 1 and 2; HRMS (ESI) [M + Na]⁺ m/z 351.1785 (calcd for C₁₇H₂₈O₆Na, 351.1784).

Cytospolide H (8): colorless oil; $R_f = 0.32$ (CHCl₃/MeOH 10:1); $[\alpha]^{20}_{D} = -88.9$ (*c* 0.01, CHCl₃); UV (CH₃CN) λ_{max} (ε) 215 (1829) nm; IR (film) $\nu_{max} = 3454$, 3360, 3197, 2924, 2853, 1736, 1662, 1462, 1440, 1375, 1242, 1176, 1107, 1065, 1023, 989 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Tables 1 and 2; HRMS (ESI) [M + Na]⁺ m/z 393.1887 (calcd for C₁₉H₃₀O₇Na, 393.1889).

Cytospolide I and J (9 and 10): colorless oil; $R_f = 0.31$ (CHCl₃/ MeOH 5:1); $[\alpha]^{20}_D = -80.3$ (*c* 0.06, CHCl₃); CD (CH₃CN, *c* 2.0 × 10⁻⁴) λ_{max} ($\Delta \varepsilon$) = 196 (-15.16) nm; UV (CH₃CN) λ_{max} (ε) 220 (2413) nm; IR (film) $\nu_{max} = 3443$, 2929, 2855, 1728, 1668, 1453, 1373, 1247, 1177, 1101, 1068, 1043, 989 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Tables 1 and 2; ESIMS [M + Na]⁺ m/z 351.13.

Cytospolide K (11): colorless oil; $R_f = 0.41$ (CHCl₃/MeOH 5:1); $[\alpha]^{20}_{D} = -40.9$ (*c* 0.07, CHCl₃); CD (CH₃CN, *c* 2.2 × 10⁻⁴) λ_{max} ($\Delta \varepsilon$) = 195 (-10.90) nm; UV (CH₃CN) λ_{max} (ε) 221 (1110) nm; IR (film) ν_{max} = 3429, 2927, 2856, 1714, 1455, 1375, 1242, 1181, 1101, 1029, 979, 916 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Table 3; HRMS (ESI) [M + Na]⁺ m/z 293.1727 (calcd for C₁₅H₂₆O₄Na, 293.1729).

Cytospolide L (12): colorless oil; $R_f = 0.50$ (CHCl₃/MeOH 10:1); $[\alpha]^{20}{}_{\rm D} = -75.0$ (c 0.01, CHCl₃); CD (CH₃CN, c 2.0 × 10⁻⁴) $\lambda_{\rm max}$ ($\Delta \varepsilon$) = 196 (-14.48) nm; UV (CH₃CN) $\lambda_{\rm max}$ (ε) 219 (1133) nm; IR (film) $\nu_{\rm max}$ = 3358, 3196, 2924, 2853, 1731, 1660, 1633, 1463, 1374, 1245, 1177, 1135, 1099 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Table 3; HRMS (ESI) [M + Na]⁺ m/z 335.1831 (calcd for C₁₇H₂₈O₅Na, 335.1834).

Cytospolide M (13): colorless crystals (ethyl acetate/petroleum ether 2:1); mp 136–137 °C; $R_f = 0.57$ (CHCl₃/MeOH 10:1);

$$\begin{split} & [\alpha]^{20}{}_{\rm D} = -25.5 \ (c \ 0.09, \ {\rm CHCl}_3); \ {\rm CD} \ ({\rm CH}_3{\rm CN}, \ c \ 5.0 \times 10^{-4}) \ \lambda_{\rm max} \\ & (\Delta\varepsilon) = 190 \ (+2.35), \ 214 \ (-3.07) \ {\rm nm}; \ {\rm CD} \ ({\rm KCl}) \ \lambda \ ({\rm mdeg}) \ 155 \ \mu {\rm g} \ {\rm of} \\ & 13 \ {\rm in} \ 250 \ {\rm mg} \ {\rm KCl} \ 215 \ (-26.06), \ {\rm positive} \ {\rm below} \ 196 \ {\rm nm}; \ {\rm UV} \\ & ({\rm CH}_3{\rm CN}) \ \lambda_{\rm max} \ (\varepsilon) \ 219 \ (1828) \ {\rm nm}; \ {\rm IR} \ ({\rm film}) \ \nu_{\rm max} = \ 3346, \ 2928, \\ & 2856, \ 1731, \ 1665, \ 1459, \ 1377, \ 1184, \ 1131, \ 1098, \ 1064, \ 992, \ 934 \ {\rm cm}^{-1}; \\ & {}^1{\rm H} \ {\rm and} \ {}^{13}{\rm C} \ {\rm NMR} \ {\rm spectroscopic} \ {\rm data}, \ {\rm see} \ {\rm Table \ 4; \ {\rm HRMS} \ ({\rm ESI}) \ [{\rm M} + \\ {\rm Na}]^+ \ m/z \ 309.1677 \ ({\rm calcd} \ {\rm for} \ {\rm Cl}_{15}{\rm H}_{26}{\rm O}_{5}{\rm Na}, \ 309.1678). \end{split}$$

Cytospolide N (14): colorless oil; $R_f = 0.71$ (CHCl₃/MeOH 10:1); $[\alpha]^{20}{}_{\rm D} = -60.0$ ($c \ 0.01$, CHCl₃); CD (CH₃CN, $c \ 5.0 \times 10^{-4}$) $\lambda_{\rm max} (\Delta \varepsilon) = 207 (-3.01)$ nm; UV (CH₃CN) $\lambda_{\rm max} (\varepsilon) 214 (1130)$ nm; IR (film) $\nu_{\rm max} = 3359$, 3205, 2925, 2854, 1733, 1661, 1633, 1462, 1371, 1241, 1179, 1099, 1059, 1028, 989, 935 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Table 4; HRMS (ESI) [M + Na]⁺ m/z 367.1732 (calcd for C₁₇H₂₈O₇Na, 367.1733).

Cytospolide O (15): colorless oil; $R_f = 0.47$ (CH₂Cl₂/*n*-hexane 20:1); $[\alpha]^{20}{}_{\rm D} = -6.3$ (*c* 0.06, CHCl₃); CD (CH₃CN, *c* 5.0 × 10⁻⁴) $\lambda_{\rm max}$ ($\Delta \varepsilon$) = 189 (-6.63), 216 (-1.83), 296 (+1.14) nm; UV (CH₃CN) $\lambda_{\rm max}$ (ε) 226 (2569), 251 (390), 278 (707) nm; IR (film) $\nu_{\rm max}$ = 3359, 3194, 2922, 2852, 1658, 1633, 1465, 1416 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Table 5; HRMS (ESI) [M + Na]⁺ m/z 291.1570 (calcd for C₁₅H₂₄O₄Na, 291.1572).

Cytospolide P (16): Colorless crystals (ethyl acetate/petroleum ether 2:1); mp 103–104 °C; $R_f = 0.42$ (CHCl₃); $[\alpha]^{20}{}_D = -105.9$ (c 0.02, CHCl₃); CD (CH₃CN, c 1.0 × 10⁻³) λ_{max} ($\Delta \varepsilon$) =198 (-1.94), 213 (-1.57), 290 (-0.88) nm; CD (KCl) λ (mdeg) 123 μ g of 16 in 250 mg KCl 289 (-16.61), 214 (-21.93); UV (CH₃CN) λ_{max} (ε) 223 (2558) nm; IR (film) ν_{max} = 3408, 2926, 2855, 1736, 1708, 1669, 1459, 1372, 1242, 1073, 1030, 987, 953, 890 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Table 5; HRMS (ESI) [M + Na]⁺ m/z 351.1787 (calcd for C₁₇H₂₈O₆Na, 351.1784).

Cytospolide Q (17): colorless crystals $(CH_2Cl_2/\text{petroleum ether 1:1})$; mp 97–98 °C; $R_f = 0.35$ (CHCl₃/MeOH 5:1); $[\alpha]^{20}_{D} = -4.4$ (*c* 0.04, CHCl₃); CD (CH₃CN, *c* 8.0 × 10⁻⁴) λ_{max} ($\Delta \varepsilon$) = 194 (-0.86), 217 (-1.48) nm; UV (CH₃CN) λ_{max} (ε) 219 (1537) nm; IR (film) ν_{max} = 3435, 3359, 2925, 2855, 1768, 1667, 1460, 1377, 1178, 1047,1023, 985 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Table 4; HRMS (ESI) [M + Na]⁺ m/z 309.1677 (calcd for C₁₅H₂₆O₅Na, 309.1678).

Decytospolide A (18): Colorless oil; $R_f = 0.48$ (CHCl₃/MeOH 10:1); $[\alpha]^{20}{}_{\rm D} = +6.1$ (*c* 0.08, CHCl₃); UV (CH₃CN) $\lambda_{\rm max}$ (*e*) 220 (823) nm; IR (film) $\nu_{\rm max} = 3431$, 2929, 2857, 1713, 1671, 1459, 1374, 1238, 1078, 1026, 989, 935 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Table 6; HRMS (ESI) [M + Na]⁺ m/z 265.1778 (calcd for C₁₄H₂₆O₃Na, 265.1780).

Decytospolide B (19:). colorless oil; $R_f = 0.51$ (CH₂Cl₂/*n*-hexane 20:1); $[\alpha]^{20}{}_{\rm D} = +26.6$ (*c* 0.02, CHCl₃); UV (CH₃CN) $\lambda_{\rm max}$ (*e*) 222 (2210), 267 (543), 276 (559) nm; IR (film) $\nu_{\rm max} = 3360, 3193, 2923, 2852, 1658, 1633, 1465, 1416, 1315, 1235, 1134, 1043, 879 cm⁻¹; ¹H and ¹³C NMR spectroscopic data, see Table 6; HRMS (ESI) [M + Na]⁺$ *m*/*z*307.1888 (calcd for C₁₆H₂₈O₄Na, 307.1885).

X-ray Crystallographic Studies of Cytospolide M (13). A colorless needle crystal of 13 ($0.05 \times 0.10 \times 0.10$ mm) was obtained by recrystallization from ethyl acetate/petroleum ether 2:1. C15H26O5 $(M_r = 286.36)$, orthorhombic, space group $P2_12_12_1$ with a = 5.3240(1)Å, b = 12.6918(3) Å, c = 23.0937(6) Å, $\alpha = \beta = \gamma = 90.00^{\circ}$, V =1560.47(6) Å³, Z = 4, $D_{calcd} = 1.219$ g/cm³. Intensity data were measured using a Cu K α radiation (graphite monochromator). A total of 4890 reflections were collected to a maximum 2θ value of 134.80° at 296 (2) K. The structure was solved by direct methods and refined by full matrix least-squares procedure. All non-hydrogen atoms were given anisotropic thermal parameters; hydrogen atoms were located from difference Fourier maps and refined at idealized positions riding on their parent atoms. The refinement converged at $R1(I > 2\sigma(I)) =$ 0.0568, wR2 = 0.1510 for 2552 independent reflections and 182 variables. CCDC-838718 (13) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam. ac.uk/data request/cif.

X-ray Crystallographic Studies of Cytospolide P (16). A colorless needle crystal of 16 $(0.10 \times 0.10 \times 0.15 \text{ mm})$ was obtained

by recrystallization from ethyl acetate/petroleum ether 2:1. C₁₇H₂₈O₆ $(M_r = 328.39)$, orthorhombic, space group $C222_1$ with a = 5.1842(1)Å, b = 22.7750(4) Å, c = 30.4741(4) Å, $\alpha = \beta = \gamma = 90.00^{\circ}$, V =3598.08(11) Å³, Z = 8, $D_{calcd} = 1.212$ g/cm³. Intensity data were measured using a Cu K α radiation (graphite monochromator). A total of 6130 reflections were collected to a maximum 2θ value of 134.92° at 296 (2) K. The structure was solved by direct methods and refined by full matrix least-squares procedure. All non-hydrogen atoms were given anisotropic thermal parameters; hydrogen atoms were located from difference Fourier maps and refined at idealized positions riding on their parent atoms. The refinement converged at $R1(I > 2\sigma(I)) =$ 0.0388, wR2 = 0.1035 for 2894 independent reflections and 210 variables. CCDC-838719 (16) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam. ac.uk/data request/cif.

X-ray Crystallographic Studies of Cytospolide Q (17). A colorless needle crystal of 17 $(0.10 \times 0.10 \times 0.12 \text{ mm})$ was obtained by recrystallization from CH₂Cl₂/petroleum ether (1:1). C₁₅H₂₆O₅ $(M_r = 286.36)$, orthorhombic, space group $P2_12_12_1$ with a = 5.1139(2)Å, b = 16.2267(5) Å, c = 19.0927(6) Å, $\alpha = \beta = \gamma = 90.00^{\circ}$, V =1584.35(9) Å³, Z = 4, D_{calcd} = 1.201 g/cm³. Intensity data were measured using a Cu K α radiation (graphite monochromator). A total of 4500 reflections were collected to a maximum 2θ value of 132.02° at 296(2) K. The structure was solved by direct methods and refined by full matrix least-squares procedure. All non-hydrogen atoms were given anisotropic thermal parameters; hydrogen atoms were located from difference Fourier maps and refined at idealized positions riding on their parent atoms. The refinement converged at $R1(I > 2\sigma(I)) =$ 0.0311, wR2 = 0.10877 for 2571 independent reflections and 185 variables. Absolute structure parameter 0.1(2). CCDC-838720 (17) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data request/cif.

Esterification of Cytospolide F (6) with (*R***)-MTPA Chloride.** Treatment of **6** (0.68 mg) with (*R*)-MTPA chloride (10 μL) in dry pyridine (0.5 mL), stirring at room temperature overnight, and purification by a mini silica gel column chromatography (500 mesh, *n*-hexane/CHCl₃, 3:1) afforded the (*S*)-MTPA ester of **6** (1.37 mg, 62%): ¹H NMR (600 MHz, CDCl₃) $\delta = 2.73$ (m, 1H, H-2), 5.57 (brs, 1H, H-3), 5.66 (dd, J = 16.2, 2.4 Hz, 1H, H-4), 5.40 (m, 1H, H-5), 2.14 (m, 1H, H-6 α), 2.14 (m, 1H, H-6 β), 1.85 (m, 1H, H-7 α), 1.78 (m, 1H, H-7 β), 4.93 (m, 1H, H-8), 4.88 (m, 1H, H-9), 1.32 (m, 1H, H-10a), 1.37 (m, 1H, H-10b), 1.37 (m, 1H, H-11a), 1.37 (m, 1H, H-11b), 1.46 (m, 1H, H-12a), 1.46 (m, 1H, H-12b), 5.02 (m, 1H, H-13), 1.27 (t, J = 6.4, 3H, H-14), 1.07 (d, J = 7.0, 3H, H-15).

Esterification of Cytospolide F (6) with (S)-MTPA Chloride: The same reaction of **6** (1.03 mg) with (S)-MTPA chloride (15 μ L) afforded the (R)-MTPA ester of **6** (2.16 mg, 64%): ¹H NMR (600 MHz, CDCl₃) δ = 2.69 (m, 1H, H-2), 5.51 (brs, 1H, H-3), 5.62 (dd, J = 16.2, 2.4 Hz, 1H, H-4), 5.12 (m, 1H, H-5), 2.21 (m, 1H, H-6 α), 2.21 (m, 1H, H-6 β), 1.92 (m, 1H, H-7 α), 1.74 (m, 1H, H-7 β), 4.74 (m, 1H, H-8), 4.84 (m, 1H, H-9), 1.14 (m, 1H, H-10a), 1.14 (m, 1H, H-10b), 1.13 (m, 1H, H-11a), 1.13 (m, 1H, H-11b), 1.36 (m, 1H, H-12a), 1.36 (m, 1H, H-12b), 4.95 (m, 1H, H-13), 1.24 (t, J = 6.4, 3H, H-14), 1.14 (d, J = 7.0, 3H, H-15).

Esterification of Cytospolide H (11) with (*R*)-MTPA Chloride. Compound 11 (0.56 mg) was treated with (*R*)-MTPA chloride according to the above procedure to afford the (*S*)-MTPA ester of 11 (0.81 mg, 55%): ¹H NMR (600 MHz, CDCl₃): δ = 2.75 (m, 1H, H-2), 5.59 (brs, 1H, H-3), 5.63 (d, *J* = 16.3 Hz, 1H, H-4), 5.36 (m, 1H, H-5), 1.90 (m, 1H, H-6 α), 2.25 (m, 1H, H-6 β), 1.85 (m, 1H, H-7 α), 1.42 (m, 1H, H-7 β), 1.45 (m, 1H, H-8 α), 1.72 (m, 1H, H-8 β), 4.66 (m, 1H, H-9), 1.52 (m, 1H, H-10a), 1.52 (m, 1H, H-10b), 1.40 (m, 1H, H-11a), 1.40 (m, 1H, H-11b), 1.67 (m, 1H, H-12a), 1.67 (m, 1H, H-12b), 5.11 (m, 1H, H-13), 1.24 (t, *J* = 6.2, 3H, H-14), 1.08 (d, *J* = 6.8, 3H, H-15).

Esterification of Cytospolide H (11) with (S)-MTPA Chloride. The same reaction of 11 (0.45 mg) with (S)-MTPA chloride afforded the (R)-MTPA ester of 11 (0.73 mg, 62%):

¹H NMR (600 MHz, CDCl₃) δ = 2.74 (m, 1H, H-2), 5.54 (brs, 1H, H-3), 5.58 (d, *J* = 16.3 Hz, 1H, H-4), 5.13 (m, 1H, H-5), 1.82 (m, 1H, H-6α), 2.14 (m, 1H, H-6β), 1.72 (m, 1H, H-7α), 1.40 (m, 1H, H-7β), 1.35 (m, 1H, H-8α), 1.64 (m, 1H, H-8β), 4.56 (m, 1H, H-9), 1.47 (m, 1H, H-10a), 1.47 (m, 1H, H-10b), 1.22 (m, 1H, H-11a), 1.22 (m, 1H, H-11b), 1.61 (m, 1H, H-12a), 1.61 (m, 1H, H-12b), 5.12 (m, 1H, H-13), 1.31 (t, *J* = 6.2, 3H, H-14), 1.17 (d, *J* = 6.8, 3H, H-15).

Esterification of Cytospolide R (18) with (*R*)-MTPA Chloride. Compound 18 (0.38 mg) was treated with (*R*)-MTPA chloride according to the above procedure to afford the (*S*)-MTPA ester of 18 (0.54 mg, 75%): ¹H NMR (600 MHz, CDCl₃) δ = 1.04 (t, *J* = 7.3, 3H, H-1), 2.47 (m, 1H, H-2a), 2.47 (m, 1H, H-2b), 2.68 (dd, *J* = 15.2, 7.7 Hz, 1H, H-4a), 2.40 (dd, *J* = 15.2, 74.9 Hz, 1H, H-4b), 3.77 (m, 1H, H-5), 1.41 (m, 1H, H-6 α), 1.78 (m, 1H, H-6 β), 2.23 (m, 1H, H-7 α), 1.48 (m, 1H, H-7 β), 4.65 (dt, *J* = 10.2, 2.4 Hz, 1H, H-8), 3.30 (t, *J* = 9.1 Hz, 1H, H-9), 1.50 (m, 1H, H-10a), 1.32 (m, 1H, H-10b), 1.16 (m, 1H, H-11a), 1.16 (m, 1H, H-11b), 1.22 (m, 1H, H-12a), 0.86 (t, *J* = 7.0, 3H, H-14).

Esterification of Cytospolide R (18) with (*S*)-MTPA Chloride. The same reaction of 18 (0.40 mg) with (*S*)-MTPA chloride afforded the (*R*)-MTPA ester of 18 (0.49 mg, 65%): ¹H NMR (600 MHz, CDCl₃) δ = 1.04 (t, *J* = 7.3, 3H, H-1), 2.47 (m, 1H, H-2a), 2.47 (m, 1H, H-2b), 2.68 (dd, *J* = 15.2, 7.7 Hz, 1H, H-4a), 2.40 (dd, *J* = 15.2, 74.9 Hz, 1H, H-4b), 3.78 (m, 1H, H-5), 1.46 (m, 1H, H-6α), 1.82 (m, 1H, H-6β), 2.28 (m, 1H, H-7α), 1.63 (m, 1H, H-7β), 4.68 (dt, *J* = 10.2, 2.4 Hz, 1H, H-8), 3.26 (dt, *J* = 9.1, 1.9 Hz, 1H, H-9), 1.30 (m, 1H, H-10a), 1.18 (m, 1H, H-10b), 1.10 (m, 1H, H-11a), 1.10 (m, 1H, H-11b), 1.20 (m, 1H, H-12a), 1.20 (m, 1H, H-12b), 1.18 (m, 1H, H-13a), 1.18 (m, 1H, H-13b), 0.83 (t, *J* = 7.0, 3H, H-14).

Acetylation of Cytospolide E (5). Treatment of 5 (0.7 mg) with Ac₂O, using the above procedure, afforded an acetate in quantitative yield, identical with the natural product 2.

Synthesis of 20 by Acetylation of Cytospolide F (6). To a solution of 6 (0.5 mg) in dry pyridine (0.5 mL) was added two drops of Ac₂O. The mixture was kept at room temperature for 16 h to afford, after usual workup, **20** quantitatively as colorless oil: $[\alpha]_{D}^{20} = -84.7$ $(c 0.04, CHCl_3); {}^{1}H NMR (600 MHz, CDCl_3) \delta = 2.75 (m, 1H, H-2),$ 5.35 (brs, 1H, H-3), 5.64 (dd, J = 15.6, 2.4 Hz, 1H, H-4), 5.52 (ddd, J = 15.6, 6.0, 4.8 Hz, 1H, H-5), 2.20 (m, 1H, H-6 α), 2.25 (m, 1H, H-6 β), 1.88 (m, 1H, H-7 α), 1.83 (m, 1H, H-7 β), 4.73 (q, J = 7.2, 1H, H-8), 4.92 (dt, J = 7.2, 3.6, 1H, H-9), 1.52 (m, 2H, H-10a, H-10b), 1.31 (m, 2H, H-11a, H-11b), 1.28 (m, 2H, H-12a, H-12b), 4.85 (m, 1H, H-13), 1.18 (t, J = 6.8, 3H, H-14), 1.18 (t, J = 6.8, 3H, H-15), 2.14 (s, 3H, 3-OAc), 2.07 (s, 3H, 8-OAc), 2.02 (s, 3H, 13-OAc); ¹³C NMR $(150 \text{ MHz}, \text{CDCl}_3) \delta = 171.7 \text{ (s, C-1)}, 45.0 \text{ (d, C-2)}, 72.6 \text{ (d, C-3)},$ 129.1 (d, C-4), 129.1 (d, C-5), 29.3 (t, C-6), 35.5 (t, C-7), 75.1 (d, C-8), 75.6 (d, C-9), 31.8 (t, C-10), 20.4 (t, C-11), 35.6 (t, C-12), 70.6 (t, C-13), 19.9 (q, C-14), 12.3 (q, C-15), 170.2 (s, 3-OAc), 20.9 (q, 3-OAc), 169.8 (s, 8-OAc), 21.1 (q, 8-OAc), 170.7 (s, 13-OAc), 21.3 (q, 13-OAc).

Synthesis of 20 by acetylation of cytospolide G (7). The same reaction was performed with 7 (0.7 mg) to afford an identical sample of 20 as a colorless oil in quantitative yield.

Synthesis of 20 by acetylation of cytospolide H (8). The same reaction was performed with 8 (0.4 mg) to afford an identical sample of 20 as a colorless oil in quantitative yield.

Synthesis of 21 by acetylation of cytospolide K (11). Treatment of **11** (0.7 mg) with Ac₂O according to the above procedure, afforded **21** quantitatively as a colorless oil. [*α*]20 D= -53.4 (*c* 0.02, CHCl₃). ¹H NMR (600 MHz, CDCl₃): δ = 2.73 (m, 1H, H-2), 5.37 (brs, 1H, H-3), 5.62 (d, *J* = 15.6 Hz, 1H, H-4), 5.49 (m, 1H, H-5), 2.206 (m, 1H, H-6*α*), 2.28 (m, 1H, H-6*β*), 1.88 (m, 1H, H-7*α*), 1.45 (m, 1H, H-7*β*), 1.50 (m, 1H, H-8*α*), 1.77 (m, 1H, H-8*β*), 4.70 (m, 1H, H-9), 1.55 (m, 1H, H-10a), 1.45 (m, 1H, H-10b), 1.30 (m, 2H, H-11a, H-11b), 1.50 (m, 2H, H-12a, H-12b), 4.87 (m, 1H, H-13), 1.19 (t, *J* = 6.6, 3H, H-14), 1.18 (t, *J* = 6.6, 3H, H-15), 2.14 (s, 3H, 3-OAc), 2.02 (s, 3H, 13-OAc); ¹³C NMR (150 MHz, CDCl₃): δ = 174.4 (s, C-1), 45.4 (d, C-2), 73.3 (d, C-3), 129.1 (d, C-4), 129.2 (d, C-5), 32.7 (t, C-6), 28.2 (t, C-7), 33.1 (t, C-8), 75.9 (d, C-9), 35.7 (t, C-10), 21.0

(t, C-11), 35.6 (t, C-12), 70.7 (t, C-13), 19.9 (q, C-14), 12.2 (q, C-15), 170.4 (s, 3-OAc), 20.9 (q, 3-OAc), 170.8 (s, 13-OAc), 21.4 (q, 13-OAc).

Synthesis of 21 by acetylation of cytospolide L (12). The same reaction was performed with 12 (0.6 mg) to afford an identical sample of 21 as a colorless oil in quantitative yield.

Acetylation of cytospolide R (18). Treatment of 18 (0.4 mg) with Ac₂O, using the above procedure, afforded an acetate in quantitative yield, identical with the natural product 19.

Cytotoxicity assay. The cytotoxic activity of tested compounds against human lung adenocarcinoma (A549), human colon cancer cells (HCT116), human hepatocarcinoma cells (QGY), human malignant melanoma cells (A375), and human leukemic cells (U937) was assayed by the MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] colorimetric method.⁵⁵ Adriamycin was used as standard compound.

Cell cycle test. A549 cells were treated with compounds **5** and **16** at 1 μ g/mL for 48 h, and QGY cells were treated with compound **16** at 10 μ g/mL for 48 h, respectively. Then the cells were trypsinized, washed with PBS twice, and then fixed with 70% ethanol on ice overnight. The fixed cells were spun down, resuspended in PBS at 1 × 10⁶ cells/mL and incubated with 50 μ g/mL propidium idide (PI) and 100 μ g/mL ribonuclease A (RNase A) at 4 °C for 30 min, before measured by flow cytometry. All experiments were performed in triplicate. The data were expressed as mean ± SD. The significance was expressed by using one-way analyses of variance (ANOVAs) of the SPSS 13.0 for Windows (SPSS Inc., Chicago, IL, USA), followed by Duncan post hoc tests. P values less than 0.05 were considered significant.

Computational section. Geometry optimizations [B3LYP/ 6-31G(d) level of theory] and TDDFT calculations were performed with Gaussian 03^{56} using various functionals (B3LYP, BH&HLYP, PBE0) and TZVP basis set. CD spectra were generated as the sum of Gaussians⁵⁷ with 3000 cm⁻¹ half-height width (corresponding to 12 at 200 nm), using dipole-velocity computed rotational strengths. Conformational searches were carried out by means of the Macromodel 9.7.211⁵⁸ software using Merck Molecular Force Field (MMFF) with implicit solvent model for chloroform. Boltzmann distributions were estimated from the ZPVE corrected B3LYP/6-31G(d) energies. The MOLEKEL⁵⁹ software package was used for visualization of the results.

ASSOCIATED CONTENT

S Supporting Information

MS, one- and two-dimensional NMR spectra for compounds 6-19, X-ray data for crystals 13, 16, and 17 (CIF), and atom coordinates and absolute energies of the computed structures. These material are available free of charge via the Internet at http://pubs.acs.org

AUTHOR INFORMATION

Corresponding Author

*E-mail: (L.L.) lilingty@126.com; (W.Z.) wenzhang1968@ 163.com.

ACKNOWLEDGMENTS

This research was financially supported by the Natural Science Foundation of China (Nos. 30873200, 41076082, 81172979), the Shanghai Pujiang Program (PJ2008), the Scientific & Technological Major Project (2009ZX09301-011). S.A. and T.K. thank the Hungarian Scientific Research Fund (OTKA, K-81701), TÁMOP 4.2.1./B-09/1/KONV-2010-0007, and National Information Infrastructure Development Institute (NIIFI 10038). We are grateful to Prof. L. Ernst and Mrs. P. H. Schulz (Technische Universität Braunschweig, Germany) and Prof. E. Hans (University of Paderborn, Germany) for the assistance with the NOEDIFF experiments. We are also indebted to

Prof. R. Riguera (Universidad de Santiago de Compostela, Spain) for his kind advice concerning the Mosher experiment.

- REFERENCES
- (1) Dräger, G.; Kirschning, A.; Thierickeb, R.; Zerlin., M. Nat. Prod. Rep. 1996, 13, 365–375.
- (2) Naves, Y. R.; Grampoloff, A. V. Helv. Chim. Acta 1942, 25, 1500–1514.
- (3) Demole, E.; Willhalm, B.; Stoll, M. Helv. Chim. Acta 1964, 47, 1152–1159.
- (4) Ishida, T.; Wada, K. J. Chem. Soc., Chem. Commun. 1975, 209–210.
- (5) Evidente, A.; Capasso, R.; Andolfi, A.; Vurro, M.; Zonno, M. C. *Phytochemistry* **1998**, *48*, 941–945.
- (6) Rivero-Cruz, J. F.; García-Aguirre, G.; Cerda-García-Rojas, C.; Mata, R. *Tetrahedron* **2000**, *56*, 5337–5344.
- (7) Yuzikhin, O.; Mitina, G.; Berestetskiy, A. J. Agric. Food Chem. 2007, 55, 7707-7711.
- (8) Evidente, A.; Cimmino, A.; Berestetskiy, A.; Andolfi, A.; Motta, A. J. Nat. Prod. 2008, 71, 1897–1901.
- (9) Rukachaisirikul, V.; Pramjit, S.; Pakawatchai, C.; Isaka, M.; Supothina, S. J. Nat. Prod. 2004, 67, 1953–1955.
- (10) Boonphong, S.; Kittakoop, P.; Isaka, M.; Pittayakhajonwut, D.; Tanticharoen, M.; Thebtaranonth, Y. J. Nat. Prod. **2001**, *64*, 965–967.
- (11) Wu, S.; Chen, Y. W.; Shao, S. C.; Wang, L. D.; Li, Z. Y.; Yang, L. Y.; Li, S. L.; Huang, R. J. Nat. Prod. 2008, 71, 731–734.
- (12) Tsuda, M.; Mugishima, T.; Komatsu, K.; Sone, T.; Tanaka, M.;
- Mikami, Y.; Kobayashi, J. J. Nat. Prod. 2003, 66, 412–415. (13) Chu, M.; Mierzwa, R.; Xu, L.; He, L.; Terracciano, J.; Patel, M.;
- Gullo, V.; Black, T.; Zhao, W.; Chan, T.; McPhail, A. T. J. Nat. Prod. 2003, 66, 1527-1530.
- (14) Bugni, T. S.; Janso, J. E.; Williamson, R. T.; Feng, X.; Bernan,
- V. S.; Greenstein, M.; Carter, G. T.; Maiese, W. M.; Ireland, C. M. J. Nat. Prod. 2004, 67, 1396–1399.
- (15) Ratnayake, A. S.; Yoshida, W. Y.; Mooberry, S. L.; Hemscheidt, T. Org. Lett. **2001**, *3*, 3479–3481.
- (16) Papendorf, O.; König, G. M.; Wright, A. D.; Chorus, I.; Oberemm, A. J. Nat. Prod. 1997, 60, 1298–1300.
- (17) Tan, Q.; Yan, X.; Lin, X.; Huang, Y.; Zheng, Z.; Song, S.; Lu, C.; Shen, Y. Helv. Chim. Acta 2007, 90, 1811–1817.
- (18) Nicoletti, R.; Lopez-Gresa, M. P.; Manzo, E.; Carella, A.; Ciavatta, M. L. *Mycopathologia* **2007**, *163*, 295–301.
- (19) Ferraz, H. M. C.; Bombonato, F. I.; Longo, L. S. Synthesis 2007, 21, 3261–3285.
- (20) Riatto, V. B.; Pilli, R. A.; Victor, M. M. *Tetrahedron* 2008, 64, 2279–2300.
- (21) Ishigami, K. Biosci. Biotechnol. Biochem. 2009, 73, 971-979.
- (22) Chowdhury, P. S.; Gupta, P.; Kumar, P. Tetrahedron Lett. 2009, 50, 7188-7190.
- (23) Mohapatra, D. K.; Sahoo, G.; Ramesh, D. K.; Rao, J. S.; Sastry,
 G. N. *Tetrahedron Lett.* 2009, 50, 5636–5639.
- (24) Krishna, P. R.; Rao, T. J. Tetrahedron Lett. 2010, 51, 4017–4019.
- (25) Yadav, J. S.; Lakshmi, K. A.; Reddy, N. M.; Prasad, A. R.; Subba Reddy, B. V. *Tetrahedron* **2010**, *66*, 334–338.
- (26) Gupta, P.; Kumar, P. Tetrahedron: Asymmetry 2007, 18, 1688–1692.
- (27) Nagaiah, K.; Sreenu, D.; R. Rao, S.; Yadav, J. S. *Tetrahedron Lett.* **2007**, *48*, 7173–7176.
- (28) Selvam, J. J. P.; Rajesh, K.; Suresh, V.; Babu, D. C.; Venkateswarlu, Y. *Tetrahedron: Asymmetry* **2009**, *20*, 1115–1119.
- (29) Kamal, A.; Reddy, P. V.; Prabhakar, S. *Tetrahedron: Asymmetry* **2009**, 20, 1120–1124.
- (30) Srihari, P.; Rao, G. M.; Rao, R. S.; Yadav, J. S. Synthesis 2010, 2407–2412.
- (31) Chakraborty, T. K.; Samanta, R.; Kumar, P. K. Tetrahedron 2009, 65, 6925-6931.
- (32) Matsuda, M.; Yamazaki, T.; Fuhshuku, K.-i.; Sugai, T. Tetrahedron 2007, 63, 8752–8760.

- (33) Ghosh, S.; Rao, R. V. Tetrahedron Lett. 2007, 48, 6937–6940.
 (34) Yadav, J. S.; Thrimurtulu, N.; Gayathri, K. U.; Reddy, B. V. S.;
 Prasad, A. R. Tetrahedron Lett. 2008, 49, 6617–6620.
- (35) Barfoot, C.; Burns, A.; Edwards, M.; Kenworthy, M.; Ahmed, M.; Shanahan, S.; Taylor, R. *Org. Lett.* **2008**, *10*, 353–356.
- (36) Chowdhury, P. S.; Gupta, P.; Kumar, P. Tetrahedron Lett. 2009, 50, 7018-7020.
- (37) Marrero, J. G.; Harwood, L. M. Tetrahedron Lett. 2009, 50, 3574–3576.
- (38) Jana, N.; Mahapatra, T.; Nanda, S. Tetrahedron: Asymmetry 2009, 20, 2622–2628.
- (39) Srihari, P.; Kumaraswamy, B.; Rao, G. M.; Yadav, J. S. *Tetrahedron: Asymmetry* **2010**, *21*, 106–111.
- (40) Prabhakar, P.; Rajaram, S.; Reddy, D. K.; Shekar, V.; Venkateswarlu, Y. *Tetrahedron: Asymmetry* **2010**, *21*, 216–221.
- (41) Srihari, P.; Kumaraswamy, B.; Bhunia, D. C.; Yadav, J. S. *Tetrahedron Lett.* **2010**, *51*, 2903–2905.
- (42) Burns, A.; McAllister, G.; Shanahan, S.; Taylor, R. Angew. Chem., Int. Ed. 2010, 49, 5574–5577.
- (43) Zhang, W.; Krohn, K.; Zia-Ullah.; Flörke, U.; Pescitelli, G.; Bari, L. D.; Antus, S.; Kurtán, T.; Rheinheimer, J.; Draeger, S.; Schulz, B. *Chem.—Eur. J.* **2008**, *14*, 4913–4923.
- (44) Zhang, W.; Krohn, K.; Egold, H.; Draeger, S.; Schulz, B. Eur. J. Org. Chem. 2008, 4320-4328.
- (45) Zhang, W.; Krohn, K.; Draeger, S.; Schulz, B. J. Nat. Prod. 2008, 71, 1078–1081.
- (46) Lu, S.; Kurtán, T.; Yang, G.; Sun, P.; Mándi, A.; Krohn, K.; Draeger, S.; Schulz, B.; Yi, Y.; Li, L.; Zhang, W. *Eur. J. Org. Chem.* **2011**, 5452–5459.
- (47) Dale, J. A.; Mosher, H. S. J. Am. Chem. Soc. 1973, 95, 512–519.
 (48) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. J. Am. Chem. Soc. 1991, 113, 4092–4096.
- (49) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. J. Org. Chem. 1991, 56, 1296–1298.
- (50) Pescitelli, G.; Kurtán, T.; Flörke, U.; Krohn, K. Chirality 2009, 21, E181–201.
- (51) Diedrich, C.; Grimme, S. J. Phys. Chem. A 2003, 107, 2524–2539.
- (52) Crawford, T. D. Theor. Chem. Acc. 2006, 115, 227-245.
- (53) Contreras, R. H.; Peralta, J. E. Prog. Nucl. Magn. Reson. Spectrosc. 2000, 37, 321–425.
- (54) Bifulco, G.; Dambruoso, P.; Gomez-Paloma, L.; Riccio, R. Chem. Rev. 2007, 107, 3744–3779.
- (55) Mosmann, T. J. Immunol. Meth. 1983, 65, 55-63.
- (56) Frisch, M. J.; Trucks, G. W.; chlegel, H. B.; Scuseria, S G. E.; Robb, M. A.; Cheeseman, J. R.; Montgomery, J. A., Jr.; Vreven, T.; Kudin, K. N.; Burant, J. C.; Millam, J. M.; Iyengar, S. S.; Tomasi, J.; Barone, V.; Mennucci, B.; Cossi, M.; Scalmani, G.; Rega, N.; Petersson, G. A.; Nakatsuji, H.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Klene, M.; Li, X.; Knox, J. E.; Hratchian, H. P.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Ayala, P. Y.; Morokuma, K.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Zakrzewski, V. G.; Dapprich, S.; Daniels, A. D.; Strain, M. C.; Farkas, O.; Malick, D. K.; Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; J. Ortiz, V.; Cui, Q.; Baboul, A. G.; Clifford, S.; Cioslowski, J.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Martin, R. L.; Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Challacombe, M.; Gill, P. M. W.; Johnson, B.; Chen, W.; Wong, M. W.; Gonzalez, C.; Pople, J. A. Gaussian 03, Revision C.02, Gaussian, Inc., Wallingford CT, 2004.
- (57) Stephens, P. J.; Harada, N. Chirality 2010, 22, 229-233.
- (58) MacroModel, Schrödinger LLC, 2009. http://www.schrodinger. com/Products/macromodel.html.
- (59) Varetto, U. MOLEKEL 5.4., 2009, Swiss National Supercomputing Centre, Manno, Switzerland.